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Chemistry 1 Introduction Qualitative analysis is a method used for identification of ions or compounds in a sample. In many cases, qualitative analysis will also involve the separation of ions or compounds in a mixture. Examples of qualitative tests would include ion precipitation reactions (solubility tests) or chemical reactivity tests. The experiments in this manual have been taken, with modifications, from the following: Laboratory Experiments: Chemistry the Central Science. Nelson, John H. and Kenneth C.

There are two general situations in which qualitative analysis is used - in the identification of a simple salt, or the identification of multiple cations in a solution. Identifying a Simple Salt The basic testing procedure for identifying a salt is as follows. Several new experiments have been added, and many questions, problems, and other material have been added or changed. Thus, instructors may alternate between the old manual and this new one. Experiments () emphasizing the principles of ionic equilibria precede those on qualitative analysis (). Ex- other texts.

Qualitative analysis is used to identify and separate cations and anions in a sample substance. Unlike quantitative analysis, which seeks to determine the quantity or amount of sample, qualitative analysis is a descriptive form of analysis. Abstract: A first-semester general chemistry laboratory experiment for college is described in which students carry out qualitative analysis to identify seven unknown colorless solutions. The experiment serves to correlate qualitative analysis with classification.

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